# organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

## Vasu,<sup>a</sup> K. A. Nirmala,<sup>b</sup> Deepak Chopra,<sup>c</sup>\* S. Mohan<sup>d</sup> and J. Saravanan<sup>e</sup>

<sup>a</sup>Vivekananda Degree College, Bangalore 560 055, Karnataka, India, <sup>b</sup>Department of Physics, Bangalore University, Bangalore 560 056, Karnataka, India, <sup>c</sup>Solid State and Structural Chemistry Unit, Indian Institute of Science, Bangalore 560 012, Karnataka, India, <sup>d</sup>PES College of Pharmacy, Hanumanthanagar, Bangalore 560 050, Karnataka, India, and <sup>e</sup>MS Ramaiah College of Pharmacy, Bangalore 560 054, Karnataka, India

Correspondence e-mail: deepak@sscu.iisc.ernet.in

#### **Key indicators**

Single-crystal X-ray study T = 293 K Mean  $\sigma$ (C–C) = 0.005 Å Disorder in main residue R factor = 0.061 wR factor = 0.208 Data-to-parameter ratio = 14.1

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# Methyl 2-(acetylamino)-4,5,6,7-tetrahydro-1-benzothiophene-3-carboxylate

The title compound,  $C_{12}H_{15}NO_3S$ , exhibits antibacterial and antifungal activities. The cyclohexene ring exhibits disorder, indicating two possible conformations of the half-chair form. The molecular structure is approximately planar, supported by an intramolecular  $N-H\cdots O$  hydrogen bond.

Received 6 August 2004 Accepted 25 August 2004 Online 31 August 2004

## Comment

Schiff bases (Csaszar & Morvay, 1983; Lakshmi *et al.*, 1985; Cohen *et al.*, 1977) and their derivatives of thiophene (El-Maghraby *et al.*, 1984; Dzhurayev *et al.*, 1992; Gewald *et al.*, 1966) possess antibacterial, antitubercular and antifungal properties. Sulfur-containing Schiff bases are most effective. The title compound, (I), shows the above-mentioned biological properties (Mohan & Saravanan, 2002, 2003).



The molecular structure of (I) is shown in Fig. 1. The deviations of atoms C5a and C6a, which constitute the portion of the cyclohexene ring of the major conformer, from the



Figure 1

The molecular structure of (I), showing 50% probability ellipsoids. Both disorder components are shown. H atoms other than H1n have been omitted. Dashed lines indicate  $N-H\cdots O$  hydrogen bonds.

© 2004 International Union of Crystallography Printed in Great Britain – all rights reserved C4/C3/C8/C7 plane are 0.412 (15) and -0.415 (15) Å, respectively. The torsion angles (Table 1) indicate that the molecule is approximately planar. An intramolecular N-H···O hydrogen bond (Table 2) forms a pseudo-sixmembered ring, thus locking the molecular conformation and eliminating conformational flexibility.

## Experimental

Cyclohexanone (3.92 g, 0.04 mol), methyl cyanoacetate (4.52 g, 0.04 mol) and sulfur (1.2 g, 0.04 mol) were mixed with ethanol (40 ml) and stirred at 315 K for 1 h with dropwise addition of diethylamine (4 ml). The product obtained was then reacted with acetic anhydride (10 ml) and heated on a water bath until the solid dissolved. The mixture was allowed to cool and the resulting solid (yield 64%) was recrystallized from ethanol.

 $D_x = 1.357 \text{ Mg m}^{-3}$ 

Cell parameters from 710

 $0.39 \times 0.30 \times 0.20 \text{ mm}$ 

Mo  $K\alpha$  radiation

reflections

 $\mu = 0.26~\mathrm{mm}^{-1}$ 

T = 293 (2) K

Block, yellow

 $\theta = 1.6\text{--}26.4^\circ$ 

#### Crystal data

C<sub>12</sub>H<sub>15</sub>NO<sub>3</sub>S  $M_r = 253.32$ Monoclinic, C2/c a = 15.289 (14) Åb = 10.644 (10) Åc = 15.790 (15) Å $\beta = 105.212 \ (13)^{\circ}$  $V = 2480 (4) \text{ Å}^3$ Z = 8

Data collection

| Bruker SMART APEX CCD area-            | 2519 independent reflections           |
|--|--|
| detector diffractometer                | 1881 reflections with $I > 2\sigma(I)$ |
| $\varphi$ and $\omega$ scans           | $R_{\rm int} = 0.039$                  |
| Absorption correction: multi-scan      | $\theta_{\rm max} = 26.4^{\circ}$      |
| (SADABS; Sheldrick, 1996)              | $h = -17 \rightarrow 18$               |
| $T_{\min} = 0.852, \ T_{\max} = 0.953$ | $k = -13 \rightarrow 7$                |
| 6489 measured reflections              | $l = -17 \rightarrow 19$               |

#### Refinement

```
Refinement on F^2
                                                      w = 1/[\sigma^2(F_o^2) + (0.1396P)^2]
R[F^2 > 2\sigma(F^2)] = 0.061
                                                           + 2.1364P
wR(F^2) = 0.208
                                                          where P = (F_o^2 + 2F_c^2)/3
S = 0.91
                                                      (\Delta/\sigma)_{\rm max} < 0.001
                                                                              -3
2519 reflections
                                                      \Delta \rho_{\rm max} = 0.32 \ {\rm e} \ {\rm \AA}
                                                      \Delta \rho_{\rm min} = -0.17 \text{ e} \text{ Å}^{-3}
179 parameters
H atoms treated by a mixture of
   independent and constrained
   refinement
```

#### Table 1

Selected torsion angles (°).

| C7-C8-C3-C2 | -178.7 (3) | O3-C11-C2-C1 | -179.8 (3) |
|-------------|------------|--------------|------------|
| C9-N1-C1-C2 | 178.7 (3)  | C1-N1-C9-C10 | -179.9 (3) |

### Table 2

Hydrogen-bonding geometry (Å, °).

| $D - H \cdots A$ | D-H      | $H \cdots A$ | $D \cdots A$ | $D - H \cdots A$ |
|------------------|----------|--------------|--------------|------------------|
| N1-H1 $n$ ···O2  | 0.83 (4) | 2.01 (4)     | 2.683 (5)    | 138 (4)          |

Atoms C5 and C6 show positional disorder and the occupancy factors of two possible sites, C5a/C6a and C5b/C6b, are 0.58 (2) and 0.42 (2), respectively. The amine H atom was located in a difference Fourier map and refined isotropically. H atoms bonded to the C atoms were positioned geometrically and allowed to ride on their parent atoms, with C-H = 0.96–0.97 Å and  $U_{iso} = 1.2U_{eq}(C)$  or  $1.5U_{eq}(C_{methyl}).$ 

Data collection: SMART (Bruker, 1998); cell refinement: SAINT (Bruker, 1998); data reduction: SAINT; program(s) used to solve structure: SIR92 (Altomare et al., 1993); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 for Windows (Farrugia, 1997) and CAMERON (Watkin et al., 1993); software used to prepare material for publication: PLATON (Spek, 2003).

The authors thank Professor T. N. Guru Row of the Indian Institute of Science, Bangalore, and the Department of Science and Technology, India, for data collection on the CCD facility set up under the IRHPA-DST program and Bangalore University. Vasu thanks Vivekananda Degree College for their support. DC thanks the CSIR for a fellowship.

### References

- Altomare, A., Cascarano, G., Giacovazzo, C. & Guagliardi, A. (1993). J. Appl. Cryst. 26, 343-350.
- Bruker (1998). SMART (Version 5.628) and SAINT (Version 6.45a). Bruker AXS Inc., Madison, Wisconsin, USA.
- Cohen, V. I., Rist, N. & Duponchel, C. (1977). J. Pharm. Sci. 66, 1322-1334. Csaszar, J. & Morvay, J. (1983). Acta Pharm. Hung. 53, 121-128.
- Dzhurayev, A. D., Karimkulov, K. M., Makhsumov, A. G. & Amanov, N.

(1992). Khim. Form. Zh. 26, 73-75.

El-Maghraby, A. A., Haroun, B. & Mohammed, N. A. (1984). Egypt. J. Pharm. Sci. 23, 327-336.

Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.

Gewald, K., Schinke, E. & Botcher, H. (1966). Chem. Ber. 99, 94-100.

Lakshmi, V. V., Sridhar, P. & Polasa, H. (1985). Indian J. Pharm. Sci. 47, 202-204

- Mohan, S. & Saravanan, J. (2002). Indian J. Heterocycl. Chem. 12, 87-88.
- Mohan, S. & Saravanan, J. (2003). Asian J. Chem. 15, 67-70.
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.

Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany. Spek, A. L. (2003). J. Appl. Cryst. 36, 7-13.

Watkin, D. M., Pearce, L. & Prout, C. K. (1993). CAMERON. Chemical Crystallography Laboratory, University of Oxford, England.